This article was downloaded by:

On: 28 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713618290

Chemical Mechanism for Fluorapatite Dissolution in Acids

Sergey V. Dorozhking^a

^a Research Institute of Fertilizers &Insectofungicides, Moscow, ussia

To cite this Article Dorozhking, Sergey V.(1996) 'Chemical Mechanism for Fluorapatite Dissolution in Acids', Phosphorus, Sulfur, and Silicon and the Related Elements, 111: 1, 4

To link to this Article: DOI: 10.1080/10426509608054633 URL: http://dx.doi.org/10.1080/10426509608054633

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Printed in Malaysia

CHEMICAL MECHANISM FOR FLUORAPATITE DISSOLUTION IN ACIDS

SERGEY V. DOROZHKIN
Research Institute of Fertilizers & Insectofungicides
Kudrinskaja sq. 1 - 155, 123242 Moscow D-242, Russia

Abstract New chemical mechanism for acidic dissolution of fluorapatite is discussed.

Modern level of knowledge about the dissolution chemistry of fluorapatite in acids is limited by chemical reaction:

$$Ca_5(PO_4)_3F + 7H_3PO_4 = 5Ca(H_2PO_4)_2 + HF$$
 (1)

But reactions having such high molecularity values are impossible. By using of various methods of solid surface state analysis (Scanning electron microscopy, electron and IR-spectroscopy, measurements surface electrokinetical potential) and chemical analysis of acid supposed solutions, Ι a new five-steps dissolution mechanism for the chemical process (1), where the each step (reaction) has the molecularity value equal 2 or 3:

$$Ca_5(PO_4)_3F + H_3O^+ = Ca_5(PO_4)_3OH + H_2F^+$$
 (2)

$$Ca_5(PO_4)_3OH + H_3O^+ = Ca_5(PO_4)_3^+ + 2H_2O$$
 (3)

$$20a_{5}(PO_{4})_{3}^{+} \triangleq 0a^{2+} + 30a_{3}(PO_{4})_{2}$$
 (4)

$$Ca_3(PO_4)_2 + 2H_3O^+ = Ca^{2+} + 2CaHPO_4 + 2H_2O$$
 (5)

$$CaHPO_4 + 2H_3O^+ = Ca^{2+} + H_3PO_4 + 2H_2O$$
 (6)

The obtained mechanism is useful for better understanding of chemical processes occurring in dental caries and artificial bones in vivo. More detailed description see^{1,2}.

REFERENCES

Dorozhkin S.V. Russ. J. Inorg. Chem., 38, 1106 (1993).
 Dorozhkin S.V. Russ. J. Inorg. Chem., 39, 229 (1994).